Raman Scattering Investigation of the Cooperative Order-Disorder Phase Transition in NH4Cl and NH4Br

Table 1. Modes which become Raman inactive upon 'cooperative disorder-order' transition in $\rm NH_4\,Cl$ and $\rm NH_4\,Br.$

$NH_4 Cl (cm^{-1})^a$	$NH_4 Br (cm^{-1})^a$
96 $(\nu_7) [Z(XX)Y]$	56 $(\nu_7) [Z(XX)Y]$
$119^{b} [Z(YZ)Y, Z(XZ)Y]$	80 [Z(YZ)Y, Z(XZ)Y]
$144^{b} \left[Z(YZ)Y, Z(XZ)Y \right]$	$141^{b} [Z(YZ)Y, Z(XZ)Y]$
	(low energy shoulder appearing
	on this peak in phase V)
$198^b [Z(YZ)Y, Z(XZ)Y]$	$180^{b} [Z(YZ)Y, Z(XZ)Y]$
$310^{c}[Z(XX)Y]$	(found at low T in phase V
	weak, broad, unresolved)

350 (ν_6) [Z(XX)Y, Z(XZ)Y, Z(XZ)Y]

^a Frequencies are given at 1 atm and 296 K. [] Indicate scattering geometry in which the mode is Raman active.

 $\frac{b}{c}$ Modes fitted to an uncoupled damped harmonic oscillator.

^c Frequency has a large uncertainty.



Fig. 4. Phase diagram for NH₄Br. Open circles with error bars represent data obtained in the present study; solid circles represent the literature values (ref. 4).

are 1.30 and 0.29 for the disordered and the ordered phase respectively. The magnitude of γ_i for the disordered phase indicates that there is a fair amount of anharmonicity associated with the librational mode. This anharmonicity probably arises from repulsive interactions between hydrogens on adjacent NH⁺₄ ions. The large frequency increase in ν_6 as the transition temperature is approached isobarically from room temperature cannot

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be accounted for with the room temperature Grüneisen constant. That is the self-energy or explicitly temperature dependent shift is larger than the volume shift.

The phase diagram⁴ for NH₄Br is shown in Fig. 4. Both II–V and V–IV transitions are disorder-order transitions with slopes of 5.0 ± 0.8 and 5 ± 1 K/kbar respectively. Experimental evidence for identifying V–IV transition as the cooperative disorder-order phase transition consists in the fact that at this transition, the Raman spectrum shows the same characteristic changes as at the disorder-order transition in NH₄Cl. This is true in the change of the number of Raman active peaks (Table 1), intensity increase of ν_5 (TO) and decrease of the Grüneisen constants upon ordering. The II–V transition is accompanied by a decrease of the Grüneisen constants, with typical changes being: 3.2 to 0.75 (ν_7); 0.72 to 0.46 ($2\nu_6$); and 0.036 to 0.003 ($\nu_2(E)$). These decreases of the γ 's suggest that there must be some type of ordering which leads to a decrease in anharmonicity. This is especially true for the librational overtone ($2\nu_6$) where the orientation-dependent part of the force constant between the NH₄⁺ ions would be expected to decrease upon ordering.

The interesting implication of the studies on NH_4Br is that both NH_4Br V and NH_4Cl II are ordered. Also, the possible existence of another high temperature phase in NH_4Cl is suggested; and this aspect is now being explored.

The present Raman data give no indication of a disorder-order transition mechanism involving a soft or other low-energy mode. However, the experimental data indicate the presence of volume- and temperature-dependent anharmonicity associated with the librational mode in the disordered state. There are also modes coupled to the translation motion of the NH_4^+ ion in the disordered phase, but their role in the phase transition is not known.

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